



JAL-003-1013022 Seat No. _____

B. Sc. (Sem. III) (CBCS) Examination

November – 2019

Biochemistry : Paper - 301

(Biomolecules)

Faculty Code : 003

Subject Code : 1013022

Time : $2\frac{1}{2}$ Hours]

[Total Marks : 70

- 1 (a) Write the correct answer for the following question. 4
- (1) Define Mutarotation.
 - (2) Define Enantiomers.
 - (3) Give the name of Non-reducing Disaccharide_____.
 - (4) Give two example of four carbon sugars.
- (b) Answer any one of the following questions : 2
- (1) What is the difference between maltose and isomaltose respect to glyosidic bond ?
 - (2) Which type of glyosidic bond present in sucrose ?
- (c) Answer any one of the following questions : 3
- (1) Write a note on derivatives of monosaccharide.
 - (2) Write a note on sugar present in milk.
- (d) Answer any one of the following questions : 5
- (1) What do you mean by optical isomer ? Explain types of isomers.
 - (2) Write a detail note on any heteropolysaccharides.
- 2 (a) Write the correct answer for the following questions : 4
- (1) Define Saponification.
 - (2) Define RM Number.
 - (3) Give the name of essential fatty acids.
 - (4) Give any two names of cholesterol derivatives.
- (b) Answer any one of the following questions : 2
- (1) Write the structure of lecithin.
 - (2) What is the biological function of phospholipids ?
- (c) Answer any one of the following questions : 3
- (1) What are the properties of triglycerols ?
 - (2) Write the structure and function of cholesterol.

- (d) Answer any one of the following questions. **5**
- (1) Write short note about the different tests to check purity of fat and oil.
 - (2) Describe the structure and function of Sphingophospholipid.
- 3** (a) Write the correct answer for the following questions. **4**
- (1) Give the name of polar amino acid with positive R group.
 - (2) Explain Biuret reaction.
 - (3) Explain Ninhydrin reaction.
 - (4) Define Isoelectric pH of Amino acid.
- (b) Answer any one of the following questions. **2**
- (1) Why amino acids acts as an ampholytes ?
 - (2) Give the name of bonds present in protein structure.
- (c) Answer any one of the following questions. **3**
- (1) Write a note on denaturation of protein.
 - (2) Write a note on Sangers method.
- (d) Answer any one of the following questions. **5**
- (1) Write a structural classification of protein.
 - (2) Write a physical property of amino acids.
- 4** (a) Write the correct answer for the following questions. **4**
- (1) Define Gene.
 - (2) Define Genome.
 - (3) Difference in Nucleotide and Nucleoside.
 - (4) Which RNA transfer amino acid to m-RNA for protein biosynthesis ?
- (b) Answer any one of the following questions. **2**
- (1) Why RNA does not obeyed Chargaff's rule ?
 - (2) Write the structure of Pyrimidine.
- (c) Answer any one of the following questions. **3**
- (1) Write a note on t-RNA.
 - (2) Write a note on central dogma.
- (d) Answer any one of the following questions. **5**
- (1) Write Griffith's experiment.
 - (2) Write the experimental evidence that proved DNA is genetic material.

- 5 (a) Write the correct answer for the following questions. 4
- (1) Liver damage is caused due to the overdose of which vitamin ?
 - (2) _____ Vitamin act as a coenzyme in carboxylation reaction.
 - (3) Give the name of precursor components for porphyrin synthesis.
 - (4) What is the importance of Beta-carotene in human diet ?
- (b) Answer any one of the following questions. 2
- (1) Write a deficiency disease of Vitamin D.
 - (2) What is Porphyrin ?
- (c) Answer any one of the following questions. 3
- (1) Describe the characteristics and function of Vitamin B₁₂.
 - (2) Give the function and deficiency condition Vitamin C.
- (d) Answer any one of the following questions. 5
- (1) Describe Vitamin K.
 - (2) Write a detailed note on break down of heme.
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